**The City School**

**North Nazimabad Boys Campus**

**Chemistry class 9**

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Q.1.choose the best ansers:

1. Oxide that is not acidic is
2. SiO2
3. P4O10
4. SO2
5. MgO
6. Amphoteric oxide among given four is
7. Na2O
8. MgO
9. Al2O3
10. SO3
11. One of four oxides that is acidic is
12. Na2O
13. MgO
14. SiO2
15. P4O10
16. K2O + H2O →
17. K(OH)3
18. KOH
19. KOH.H2O
20. KO + H2 + O2
21. Non-metal oxides showing no reaction with acids or alkalis are called as
22. Amphoteric acids
23. neutral oxides
24. acidic oxides
25. basic oxides
26. What are products of given reaction? CO2 + H2O →
27. CO
28. H3O
29. H2CO3
30. CO2 + OH-
31. Acidic oxides neutralize
32. bases
33. alkalis
34. salt solutions
35. Both A and B
36. CaCO3 + HCl →
37. Ca(OH)2 + H2O
38. CaCl2 +H2O
39. CaCl2 + H2 + C(s)
40. CaCl2 + H2O + CO2

Q.2.Give some examples of neutralization from daily life: