## The City School

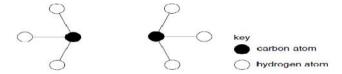
## North Nazimabad Boys Campus

Chemistry test (alkanes and alkenes)
Class:11



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1. The diagram shows a model of an organic compound



What is the name of this compound?

A ethane

B ethanoic acid

C ethanol

**D** ethene

2. Bitumen is a substance obtained from the fractional distillation of petroleum. What are the boiling points and the sizes of the molecules in bitumen?

	boiling points	sizes of molecules
Α	high	large
В	high	small
С	low	large
D	low	small

3. The structures of three compounds are shown.

Why do these substances all belong to the same homologous series?

- A They all contain an even number of carbon atoms.
- B They all contain the same functional group.
- C They are all hydrocarbons.
- D They are all saturated.
- 4. Which column describes ethane and which column describes ethene?

	hydrocarbon			
	1	2	3	4
state at room temperature	gas	gas	liquid	liquid
reaction with oxygen	burns	burns	burns	burns
reaction with aqueous bromine	no reaction	decolourises bromine	no reaction	decolourises bromine

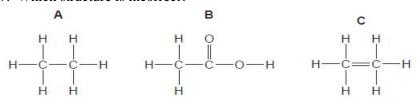
- A 1 (ethane) and 2 (ethene)
- B 1 (ethane) and 3 (ethene)
- C 2 (ethene) and 3 (ethane)
- D 3 (ethane) and 4 (ethene)
- 5. Which properties do butane, propene and ethanol all have?

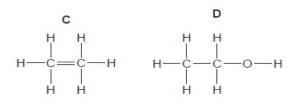
	burn	polymerise
A	1	1
В	✓	x
С	×	✓
D	x	x

6. When a suitable catalyst is used, ethene reacts with steam.

What is the structure of the compound formed?

7. Which structure is incorrect?





8.

Eth ane is a me mbe

r of the hydrocarbon series with the general formula

- A CnH2n+2
- C CnH2n-n
- B CnH2n
- D CnH2n-6
- 9. Given the equation:

$$C_{11}H_{94} \xrightarrow{450^{\circ}C} C_{5}H_{10} + C_{4}H_{8} + C_{2}H_{4} + H_{2}$$

Which type of reaction does this equation represent?

- A addition
- C hydrogenation
- B cracking
- D substitution

## 10. Which is an isomer of butane

